Phytochemical and Larvicidal Studies on Stemona curtisii: Structure of a New Pyrido[1,2-a]azepine Stemona Alkaloid

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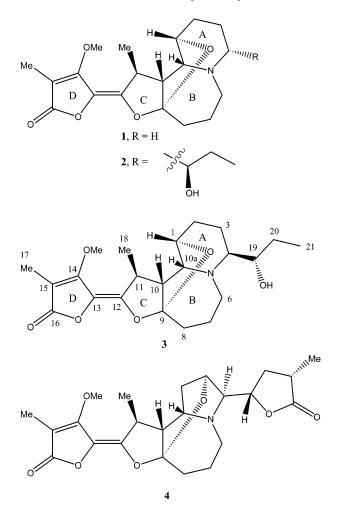
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A new pentacyclic *Stemona* alkaloid, stemocurtisinol (3), with a pyrido[1,2-*a*]azepine A,B-ring system, and the known pyrrolo[1,2-a]azepine alkaloid oxyprotostemonine (4) have been isolated from a root extract of S. curtisii. The structure and relative stereochemistry of stemocurtisinol was determined by spectral data interpretation and X-ray crystallography. This compound is a diastereoisomer of oxystemokerrin and has the opposite configuration at C-4 and C-19. The individual alkaloid components showed significant larvicidal activity (IC₅₀ 4-39 ppm) on mosquito larvae (*Anopheles minimus* HO).

The Stemona group of alkaloids includes more than 40 different natural products that have been structurally classified into five different groups.¹ The pyrrolo[1,2-a]azepine (5,7-bicyclic A,B-ring system) nucleus is common to all compounds in these groups. In 2003 we reported the structure of stemocurtisine (1), the first example of a Stemona alkaloid with a pyrido[1,2-a]azepine A,B-ring system (that is, a 6,7-bicyclic A,B-ring system), isolated from the roots of *S. curtisii* Hook. Later in that year Hofer and Greger reported the isolation of five Stemona alkaloids with the pyrido[1,2-*a*]azepine A,B-ring system, including stemocurtisine (1), which they named pyridostemin, and oxystemokerrin (2) from an unidentified *Stemona* species (HG 915) and S. kerrii, respectively.³ Interestingly, these workers also examined the phytochemicals from *S. curtisii*; the major and minor components were stemofoline, a known pyrrolo[1,2-a]azepine Stemona alkaloid, and 2'hydroxystemofoline, respectively, with trace amounts of oxystemokerrin (2) also detected. Our root samples of S. curtisii were collected in Trang Province in southern Thailand in May 2002, and a voucher specimen was deposited in the Herbarium at Chiang Mai University. In turn the S. curtisii plant samples of Hofer and Greger were collected from Sutun Province in southern Thailand (date not disclosed). Unfortunately, no voucher plant specimen from the collection of Hofer and Greger was deposited in Thailand, making a botanical comparison difficult. However, we are confident that Mr. J. Maxwell, a taxonomist from the Herbarium at Chiang Mai University and an expert of tropical plants, has identified the correct plant species by comparing our specimen with the key to the genera (Stemona) of Duyfjes⁴ and Prain.⁵ We report here the isolation of two further Stemona alkaloids from the roots of this plant and the larvicidal activity of the crude root extract of S. curtisii, and its three main alkaloid components, on mosquito larvae (Anopheles minimus HO).

Preparative TLC of the crude ethanol extracts of the dried root (400 g) of S. cutisii gave stemocurtisine (1, 20.8 mg), stemocurtisinol (3, 12.7 mg), and oxyprotostemonine (4, 10.5 mg). The latter alkaloid was isolated in trace amounts from the roots of S. kerrii and S. curtisii by Hofer and Greger,3 and the 1H and 13C NMR and MS data of 4 were almost identical to that reported by these workers.



Compound **3** was obtained as pale yellow needles (mp 209-211 °C) by careful and slow evaporation of a solution

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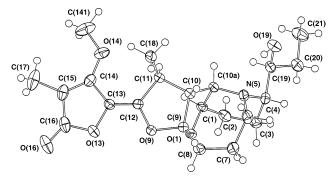


Figure 1. Molecular projection of 3, showing 50% probability amplitude displacement ellipsoids. Hydrogen atoms have arbitrary radii of 0.1 Å.

of **3** in ethyl acetate. HRMS (EI +ve, m/z [M⁺] 405.2100, calcd 405.2151) indicated that 3 had the molecular formula C₂₂H₃₁NO₆. The ¹H and ¹³C NMR spectra of 3 indicated the presence of the ABCD-ring system of stemocurtisine (1), including the ether bridging structure between C-1 and C-9 and the C-4, 1-hydroxypropyl A-ring side chain. NOESY experiments showed significant cross-peaks between H-19 and H-10a and H-19 and H-2 β , revealing the β -configuration (axial orientation) of the C-4 substituent relative to the axial protons H-2 β and H-10a (Figure 1). X-ray structural analysis confirmed the molecular formula of 3 and revealed its connectivity and relative stereochemistry (Figure 1). In the crystal, the hydroxyl hydrogen, H(19O), is hydrogen-bonded intramolecularly to N(5) at a distance of 2.04(3) Å (Figure 1). The absolute configuration of 3 was not established but is assumed, based on the known configurations of Stemona alkaloids with similar C,D-ring structures.^{6,7} The NMR data of 2 and 3 were significantly different, especially in their ¹³C NMR chemical shifts for the carbons near the C-4, 1-hydroxypropyl side chain, consistent with these compounds being epimeric at C-4. Significant differences were observed in the chemical shifts for C-6 (δ 42.8 in 2³ and δ 54.8 in 3) and C-19 (δ 70.5 in 2^3 and δ 67.9 in 3). The differences in the C-6 chemical shifts are consistent with the C-4 substituent having an equatorial disposition in 2 and an axial disposition in **3**. Indeed the ¹³C NMR chemical shifts for C-6 in $\mathbf{1}^2$ and 3 were almost identical, whereas that in 2 is about 12 ppm upfield due to the γ -gauche effect⁸ of the C-4 substituent on C-6. Compounds 2 and 3 also have opposite configurations at the carbinol carbon C-19.

The full ¹H and ¹³C NMR spectral assignments for **3**, based on extensive COSY, TOCSY, NOESY, HMQC, and HMBC experiments, are shown in Table 1. As described earlier, NOESY experiments were used to determine the relative α - or β -orientation of the protons.²

The larvicidal activity of the crude root extract of *S. curtisii* and that of compounds **1**, **3**, and **4** on mosquito larvae (*Anopheles minimus* HO), using the WHO method to determine the LC_{50} ,⁹ are shown in Table 2. While the crude ethanol extract showed a LC_{50} of 81 ppm, the individual alkaloid components were significantly more potent (LC_{50} of 4–39 ppm). The most potent was oxyprotostemonine **4**, having a LC_{50} of 4 ppm.

In conclusion, two further pentacyclic *Stemona* alkaloids, stemocurtisinol (**3**) and oxyprotostemonine (**4**), have been isolated from a root extract of *S. curtisii*. Stemocurtisinol (**3**) is a diastereoisomer of oxystemokerrin (**2**) and has the opposite configuration at C-4 and C-19. The individual alkaloid components showed significant larvicidal activity on mosquito larvae (*Anopheles minimus* HO). The differences between the phytochemical profiles of *S. curtisii*

 Table 1.
 1³C NMR (75 MHz) and ¹H NMR (500 Mz) Spectral Data of 3 in CDCl₃ Solution

position	$\delta_{\rm C}$ (DEPT)	$\delta_{\rm H}$ (mult., <i>J</i> (Hz), assign.)	HMBC	
<u>.</u> 1	75.4 (CH)	4.05 (s)	H-2α, H-2β, H-3α,	
1	75.4 (CII)	4.03 (3)	H-3 β , H-10, H-10a	
2	22.4 (CH ₂)	1.73 (dd, 5.8,	H-3α, H-4	
~	22.4 (0112)	12.3, β)	11 000, 11 4	
		1.95 (m, α)		
3	18.4 (CH ₂)	1.36 (m, β)	H-1, H-2β, H-4, H-19	
	()	1.96 (m, α)	,,,,	
4	65.5 (CH)	2.53 (m)	H-3 α , H-3 β , H-6 α ,	
	. ,	. ,	H-6β, H-19	
6	54.8 (CH ₂)	2.92 (dd, 4.5,	H-4, H-7 α , H-7 β ,	
		15.5, α)	H-8 α , H-8 β , H-10a	
		3.48 (m, β)		
7	25.8 (CH ₂)	1.65 (m, β)	H-6α, H-6β, H-8α,	
			H-8 β	
		1.99 (m, α)		
8	33.5 (CH ₂)	1.76 (dd, 5.8,	H-6α, H-6β, H-7α,	
		13, β)	H-7β, H-10	
		2.36 (dd, 4.1,		
		13, α)		
9	120.1 (C)		H-7 α , H-7 β , H-8 α , H-8 β	
10	50 0 (CII)	0 70 (1 4 7)	H-10, H-10a, H-11	
10	56.9 (CH)	2.70 (d, 4.7)	H-10a, H-11, Me (18)	
10a	57.5 (CH)	3.40 (s)	H-2 α , H-2 β , H-10, H-11	
11	39.3 (CH)	3.07 (quin, 6.1)	H-10, H-10a,	
12	140.9 (C)		Me (18)	
12	146.8 (C)		H-10, H-11, Me (17),	
13	125.0 (C)		Me (18) H-11, Me (17)	
13	162.7 (C)		OMe, Me (17)	
14	97.5 (C)		Me (17)	
16	169.7 (C)		Me (17) Me (17)	
10	9.2 (CH ₃)	2.07 (s)	OMe	
18	22.6 (CH ₃)	1.38 (d, 7)	H-10, H-11	
19	67.9 (CH)	3.50 (m)	H-4, H-20, Me (21)	
20	26.4 (CH ₂)	1.25 (m)	H-4, H-19, Me (21)	
		1.60 (m)	,0, (~-)	
21	10.3 (CH ₃)	1.02 (t, 7.3)	H-19, H-20	
OMe	58.9 (CH ₃)	4.15 (s)	, -	

Table 2. Larvicidal Activity of *S. curtisii* on Mosquito Larvae

 (Anopheles minimus HO) Using the WHO Method⁹

extract/compound	LC ₅₀ (ppm)
ethanol crude extract	81
1	18
3	39
4	4

studied by us and by Hofer and Gerber may be due in part to the fact that these plants were harvested from different geographic regions and possibly at different seasons.¹⁰

Experimental Section

General Experimental Procedures. As described previously.²

Plant Material. The plant material [a voucher specimen is deposited at the Herbarium (number 17581) of the Department of Biology, Chiang Mai University] was collected as described previously.²

Extraction and Isolation. The crude alkaloid material (0.9 g) that was described previously² was chromatographed on silica gel (100 mL) using gradient elution from 100% dichloromethane to 50% methanol-dichloromethane containing 1% concentrated aqueous ammonia as eluent. A total of 1500 mL of eluent was collected in test tubes of 25 mL. On the basis of TLC analysis these fractions were pooled to give 18 fractions. Fractions 12 (62.2 mg) and 13 (86.2 mg) were combined and rechromatographed by preparative TLC (dichloromethane-methanol-aqueous ammonia, 94:5:1) to give 20.8 mg of pure stemocurtisine (1). Fraction 10 (136.8 mg) was chromatographed on silica gel using gradient elution from 100%

dichloromethane to 50% methanol-dichloromethane containing 1% concentrated aqueous ammonia as eluent; six fractions were collected. Fraction 2 (65.8 mg) and fraction 3 (30.6 mg) were rechromatographed by preparative TLC (dichloromethane-methanol-ammonia, 96:4:1) to give 12.7 mg of pure stemocurtisinol (3) and 10.5 mg of pure oxyprotostemonine (4), respectively. The ¹H and ¹³C NMR data of 4 were identical to those reported in the literature.³

Stemocurtisinol (3): pale yellow needles (ethyl acetate); mp 209–211 °C; $[\alpha]_D^{25}$ +233° (c 0.334, CHCl₃); IR (film) ν_{max} 2930, 1746, 1692, 1621, 1461, 1397, 1367, 1286, 1216, 1155, 1025, 994.62, 754 cm⁻¹; ¹H and ¹³C NMR, see Table 1; HREIMS m/z 405.2100 [M]⁺, calcd for C₂₂H₃₁NO₆ 405.2151.

X-ray Structure Determination of 3: $C_{22}H_{31}NO_6$, M =405.5, orthorhombic, space group $P2_12_12_1$ (D_2^4 , No. 19), a =7.708(3) Å, b = 14.177(6) Å, c = 18.972(8) Å, V = 2073 Å³, D_c (Z = 4) = 1.299 g cm⁻³, $\mu_{M0} = 0.09$ mm⁻¹; specimen 0.45 \times 0.15×0.13 mm; $T_{\min/\max}$ (multiscan correction) = 0.90; $2\theta_{\max}$ = 58°; N_{total} (full sphere) = 26 220, merging to N(unique) = 3015 ($R_{\text{int}} = 0.035$), $N_{\text{obs}}(F > 4\sigma(F)) = 2631$; R = 0.035, $R_{\text{w}} =$ 0.041 (weights $(\sigma^2(F) + 0.0003F^2)^{-1}$). $(x, y, z, U_{iso})_H$ refined. $|\Delta \rho_{\text{max}}| = 0.22(2)$ e Å⁻³. Bruker AXS instrument, *T* ca. 153 K; monochromatic Mo K α radiation, $\lambda = 0.71073$ Å. CCDC #226011.

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